

Littérature associée. Structure cristalline du *N*-(4,6-diméthylpyridin-2-yl)benzamide (Rodier, Piessard, Le Baut & Michelet, 1986). Structure cristalline du *N*-éthyl-*N*-(4,6-pyridin-2-yl)benzamide (Rodier, Piessard, Le Baut & Brion, 1987).

Références

B. A. FRENZ & ASSOCIATES, INC. (1982). *SDP Structure Determination Package*. College Station, Texas, EU.

JOHNSON, C. K. (1976). *ORTEP*. Rapport ORNL-5138. Oak Ridge National Laboratory, Tennessee, EU.

MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J.-P. & WOOLFSON, M. M. (1982). *MULTAN11/82. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Univ. de York, Angleterre, et de Louvain, Belgique.

RODIER, N., PLESSARD, S., LE BAUT, G. & BRION, J.-D. (1987). *Bull. Soc. Chim. Fr.* pp. 250–254.

RODIER, N., PLESSARD, S., LE BAUT, G. & MICHELET, A. (1986). *Bull. Soc. Chim. Fr.* pp. 418–422.

Acta Cryst. (1992). **C48**, 558–559

Structure of *N,N*-Bis(2,3,4,5,6-Pentafluorophenyl)oxamide

BY KENTARO YAMAGUCHI AND GO MATSUMURA

School of Pharmaceutical Sciences, Showa University, 1-5-8 Hatanodai, Shinagawa-ku, Tokyo 142, Japan

AND NAOKI HAGA AND KOICHI SHUDO

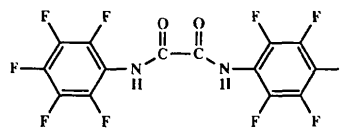
Faculty of Pharmaceutical Sciences, University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113, Japan

(Received 18 June 1991; accepted 12 August 1991)

Abstract. C₁₄H₂F₁₀N₂O₂, *M_r* = 420.16, monoclinic, *P*2₁/*n*, *a* = 14.206 (1), *b* = 4.990 (1), *c* = 10.964 (3) Å, β = 111.68 (1)°, *V* = 722.3 (1) Å³, *Z* = 2, *D_x* = 1.932 Mg m⁻³, λ(Cu Kα₁) = 1.54050 Å, μ = 1.974 mm⁻¹, *F*(000) = 412, *T* = 295 K, final *R* = 0.040 for 971 reflections. The midpoint of the oxamide carbon chain lies on a crystallographic inversion center. An intermolecular hydrogen bond, N—H⋯O [N(1)—O(3) 2.877 (2) Å], was observed.

Experimental. (I) was obtained by the reaction of pentafluoroaniline with oxalyl chloride in C₆H₆. Recrystallization from CH₃(CH₂)₄CH₃/CH₃COO-C₂H₅ gave colorless prisms (0.30 × 0.30 × 0.45 mm) having m.p. 515.5–517.0 K. The combustion analysis of this compound was consistent with the structure. Rigaku AFC-5 four-circle diffractometer used with ω-2θ-scan method, ω-scan width (1.3 + 0.14 tan θ)° and scan speed 16° min⁻¹. Lattice parameters obtained from least-squares analysis of 20 reflections with 2θ values ranging from 55 to 61°. Of 1323 reflections scanned within index range *h* -15–15, *k* 0–5, *l* 0–12 up to sin θ/λ < 0.56 Å⁻¹, *R*_{int} = 0.018 for 69 reflections, 1072 unique reflections [*F* > σ(*F*)] classified as observed. Three standard reflections measured every 150 reflections, intensity variation < 3%. Intensities corrected for Lorentz and polarization factors, but absorption correction not applied. Structure solved using program package *SAP85*

(Yao, Zheng, Qian, Han, Gu & Fan, 1985), a version of *MULTAN80* (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980). The refinement was carried out by the full-matrix least-squares method with anisotropic temperature factors for non-H atoms. The function minimized was $\sum w[(|F_o|)^2 - (|F_c|)^2]^2$ with $w = 1/[\sigma^2(F_o) + 0.016(F_o)]$; σ(*F_o*) was determined from counting statistics. All H atoms located from a difference map and theoretical calculations were refined, initial thermal parameters set at equivalent isotropic thermal parameters of each bonded atom. Final discrepancy indices *R* = 0.040, *wR* = 0.042, *S* = 2.018 for 136 variables and 971 reflections with *F* > 3σ(*F*). Maximum Δ/σ = 0.27 in final least-squares cycle. Final difference Fourier excursions 0.19 and -0.23 e Å⁻³. All major computations performed on a PANAFACOM computer with the *RCRYSTAN* (Rigaku Corporation, 1985) X-ray analysis program system. The atomic scattering factors were taken from *International Tables for X-ray Crystallography* (1974, Vol. IV).



(I)

Table 1. Atomic coordinates and equivalent isotropic thermal parameters (\AA^2)
$$B_{\text{eq}} = \frac{1}{3} \sum_i \sum_j B_{ij} a_i^* a_j^* a_i a_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	B_{eq} (\AA^2)
N(1)	0.9503 (1)	0.3442 (3)	0.3423 (1)	3.15 (4)
C(2)	0.9736 (1)	0.5469 (3)	0.4289 (1)	3.29 (5)
O(3)	0.9561 (1)	0.7815 (2)	0.4018 (1)	4.59 (4)
C(4)	0.9003 (1)	0.3854 (3)	0.2065 (1)	3.02 (5)
C(5)	0.9328 (1)	0.5723 (4)	0.1375 (1)	3.64 (5)
C(6)	0.8804 (1)	0.6168 (4)	0.0058 (1)	4.18 (6)
C(7)	0.7962 (1)	0.4688 (4)	-0.0612 (1)	4.07 (6)
C(8)	0.7652 (1)	0.2749 (4)	0.0035 (1)	3.81 (6)
C(9)	0.8157 (1)	0.2373 (3)	0.1362 (1)	3.28 (5)
F(10)	1.0184 (1)	0.7110 (2)	0.1966 (1)	5.45 (4)
F(11)	0.9131 (1)	0.8025 (3)	-0.0576 (1)	6.32 (5)
F(12)	0.7454 (1)	0.5122 (3)	-0.1887 (1)	6.16 (4)
F(13)	0.6848 (1)	0.1209 (3)	-0.0618 (1)	5.79 (4)
F(14)	0.7816 (1)	0.0533 (2)	0.1977 (1)	4.46 (3)

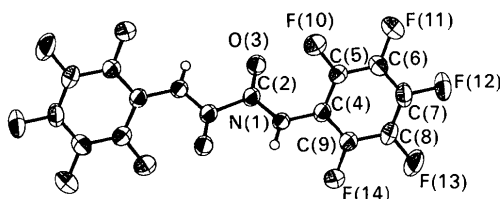


Fig. 1. ORTEP drawing of (I). Ellipsoids are drawn at the 50% probability level while isotropic H thermal parameters are represented by spheres of arbitrary size.

Final atomic parameters are listed in Table 1.* Bond lengths and angles are listed in Table 2. Fig. 1 shows an ORTEP drawing (Johnson, 1976) of the molecule with its atom labels.

* Tables of H-atom coordinates, anisotropic thermal parameters and structure-factor amplitudes have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54496 (5 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: AS0521]

Acta Cryst. (1992). C48, 559–561

Structure of *N*-(2,3,5,6-Tetrafluoropyridyl)-*N'*-phenylurea

BY KENTARO YAMAGUCHI AND GO MATSUMURA

School of Pharmaceutical Sciences, Showa University, 1-5-8 Hatanodai, Shinagawa-ku, Tokyo 142, Japan

AND NAOKI HAGA AND KOICHI SHUDO

Faculty of Pharmaceutical Sciences, University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113, Japan

(Received 18 June 1991; accepted 12 August 1991)

Abstract. $\text{C}_{12}\text{H}_7\text{F}_4\text{N}_3\text{O}$, $M_r = 285.20$, monoclinic, $P2_1/n$, $a = 7.594$ (1), $b = 24.316$ (4), $c = 6.298$ (4) \AA , $\beta = 96.47$ (1) $^\circ$, $V = 1155.7$ (3) \AA^3 , $Z = 4$, $D_x =$

Table 2. Bond lengths (\AA) and angles ($^\circ$)

N(1)—C(2)	1.342 (2)	N(1)—C(4)	1.408 (2)
C(2)—O(3)	1.211 (2)	C(4)—C(5)	1.383 (3)
C(5)—F(10)	1.339 (2)	C(5)—C(6)	1.376 (2)
C(6)—F(11)	1.340 (2)	C(6)—C(7)	1.366 (2)
C(7)—F(12)	1.333 (2)	C(7)—C(8)	1.365 (3)
C(8)—F(13)	1.343 (2)	C(8)—C(9)	1.377 (2)
C(9)—F(14)	1.333 (2)	C(9)—C(4)	1.378 (2)
C(2)—C(2')	1.533 (4)		
C(2)—N(1)—C(4)	122.3 (1)	O(3)—C(2)—N(1)	125.5 (1)
C(9)—C(4)—C(5)	117.0 (1)	C(9)—C(4)—N(1)	120.5 (1)
C(5)—C(4)—N(1)	122.5 (1)	F(10)—C(5)—C(6)	117.8 (1)
F(10)—C(5)—C(4)	120.7 (1)	C(6)—C(5)—C(4)	121.4 (1)
F(11)—C(6)—C(7)	119.8 (1)	F(11)—C(6)—C(5)	119.9 (1)
C(7)—C(6)—C(5)	120.3 (2)	F(12)—C(7)—C(8)	120.5 (1)
F(12)—C(7)—C(6)	120.2 (2)	C(8)—C(7)—C(6)	119.3 (1)
F(13)—C(8)—C(7)	120.2 (1)	F(13)—C(8)—C(9)	119.6 (2)
C(7)—C(8)—C(9)	120.2 (1)	F(14)—C(9)—C(8)	119.0 (1)
F(14)—C(9)—C(4)	119.4 (1)	C(8)—C(9)—C(4)	121.6 (1)
N(1)—C(2)—C(2')	112.9 (2)	O(3)—C(2)—C(2')	121.5 (2)

Symmetry code: (i) $2 - x, 1 - y, 1 - z$.

Related literature. The crystallographic studies of several urea cytokinins have been carried out in order to provide the basis for consideration of the stereochemical structure–activity relationships (Yamaguchi & Shudo, 1991).

References

- JOHNSON, C. K. (1976). *ORTEP*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCO, J.-P. & WOOLFSON, M. M. (1980). *MULTAN80. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Univs. of York, England, and Louvain, Belgium.
- Rigaku Corporation (1985). *RCRYSTAN. X-ray Analysis Program System*. Rigaku Corporation, Tokyo, Japan.
- YAMAGUCHI, K. & SHUDO, K. (1991). *J. Agric. Food Chem.* **39**, 793–796.
- YAO, J.-X., ZHENG, C.-D., QIAN, J.-Z., HAN, F.-S., GU, Y.-X. & FAN, H.-F. (1985). *SAP85. A Computer Program for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Institute of Physics, Chinese Academy of Sciences, Beijing, People's Republic of China.

1.639 Mg m^{-3} , $\lambda(\text{Cu } K\alpha_1) = 1.54050 \text{ \AA}$, $\mu = 1.322 \text{ mm}^{-1}$, $F(000) = 576$, $T = 295 \text{ K}$, final $R = 0.055$ for 1194 reflections. The phenyl ring lies at